Sn Etching With Hydrogen Plasma

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- Introduction
- Sn Film Growth
- Hydrogen Cleaning Process
- Experimental Setup
- Results and Discussion
- Conclusion

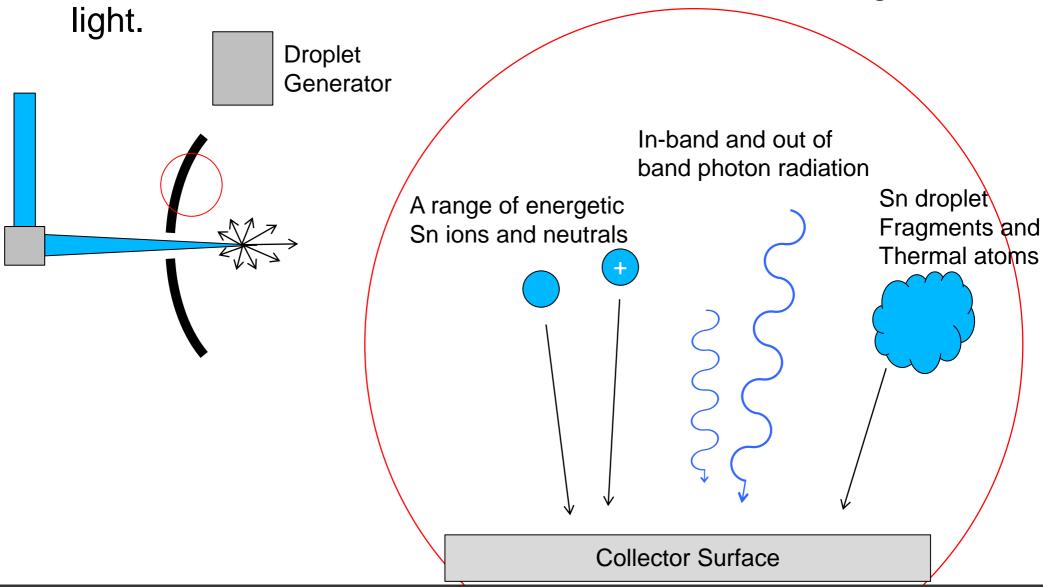
Introduction

- Sn deposition on EUV collector optics cripples reflectivity with just a couple nanometers of deposition.
- Thermal evaporation is not a solution, high energy ion sputtering is not a solution. Other methods required.
 - High temperature would result in multilayer diffusion.
 - Sputter will remove multilayers as well as the Sn.
- One possible solution is the use of hydrogen plasma to create volatile SnH₄ to remove Sn from the surface of the collector.
- The effects of pressure, flow rate, surface size on Sn removal rates are examined.

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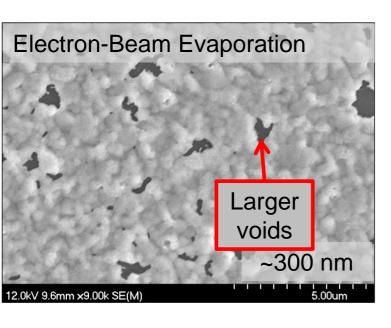
Flux Incident on Collector Optics

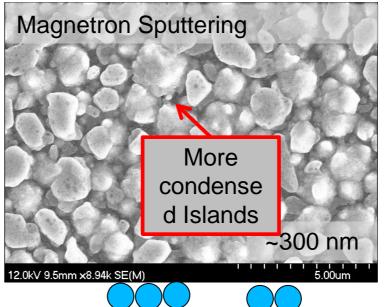
Sn comes from the condensable fuel used in creating EUV light.



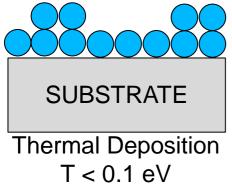
Sn Film Growth

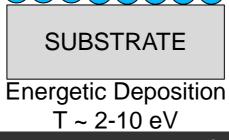
- Sn growth depends on adatom-surface interaction.
- Sn-Si and Sn-Ru growth exhibits Stranski-Krastanov growth:
 1-2 ML of coverage with addition island growth.





- Sn-Sn:Sn-Si → 1.35
- 19.5% mismatch resides slightly within SK growth¹
- Variations in deposition parameters can affect how Sn is grown.

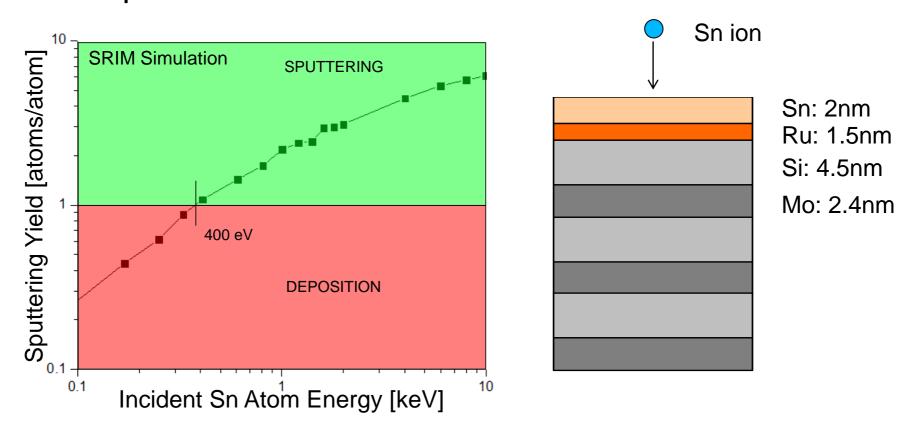




1) Summary Abstract: Interface formation in IV--IV heterostructures: Tin on silicon M. Zinke-Allmang, H.-J. Gossmann, L. C. Feldman, and G. J. Fisanick, J. Vac. Sci. Technol. A 5, 2030 (1987), DOI:10.1116/1.574912

Deposition of Sn from EUV Plasma

 Energetic sputtering of deposited Sn by non-mitigated EUV plasma species is insufficient below 400 eV.

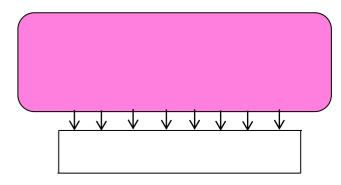


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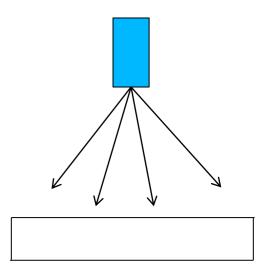
Hydrogen Plasma Vs. Atomic Hydrogen Source⁹

- ~10x Higher Sn removal rates with H₂ plasma source versus atomic H* source in same tool configuration
- Increased uniformity
 - Plasma is a uniform radical source across the Sn removal plane
 - Located above Sn contamination
 - Radicals readily created within proximal distance of Sn contamination.
 - Atomic hydrogen sources are point sources
 - 1/r² flux drop off
 - Relies on long distance diffusion

H₂ Plasma Source



Atomic H* Source



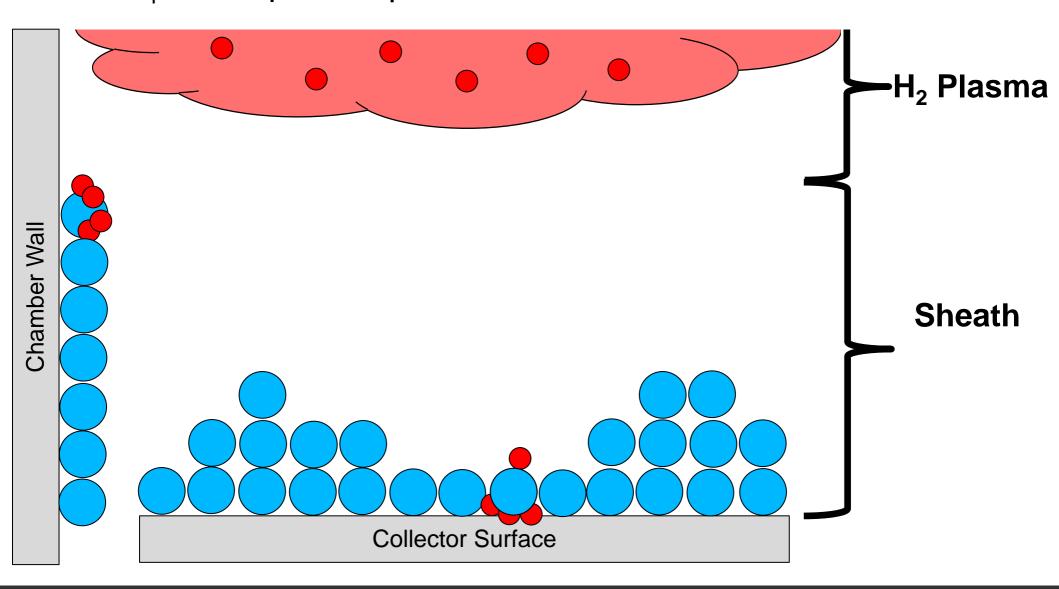
How Hydrogen Plasma Cleans

A Stoliantia and the surface.

H₂ Plasma H⁺ lons and H* **Radicals Chamber Wall Sheath** Collector Surface

Concerns

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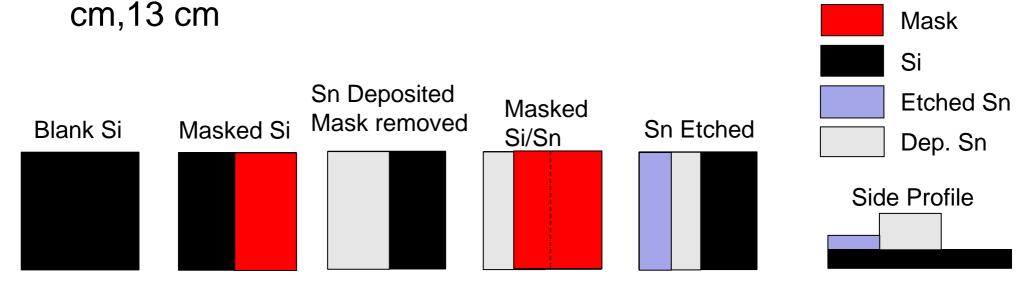


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Experimental Setup

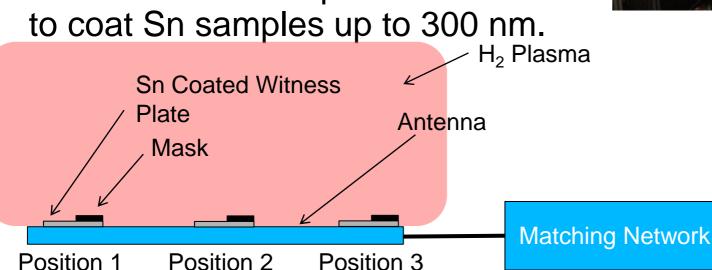
- Large stainless steel plate (16 cm wide) attached to a 13.56 MHz, 500 W max RF-generator. Samples etched 20 min.
- Pressures explored at 100 sccm for 80, 300, 500, 1000 mTorr using a gate valve to adjust pump rate.
- Flow rates of 50, 100, 500, 950 sccm explored at 80 mTorr.

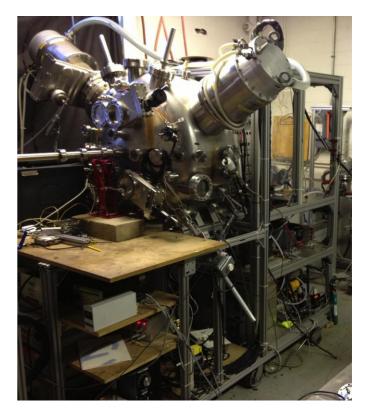
 Masked, Sn coated Si witness plates located at three different locations from the center of the RF-antenna plate: 5 cm, 9



Experimental Setup

- RF-antenna installed in CPMI's XTS 13-35 source chamber to allow pressure/flow rate manipulation with adequate chamber size.
- Langmuir triple probes are utilized to measure plasma parameters at positions 1-3.
- Electron-beam evaporation utilized



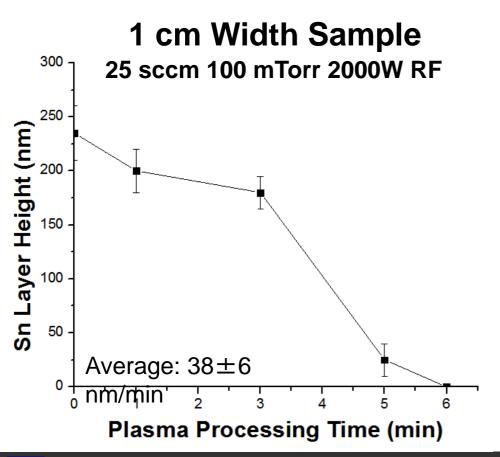


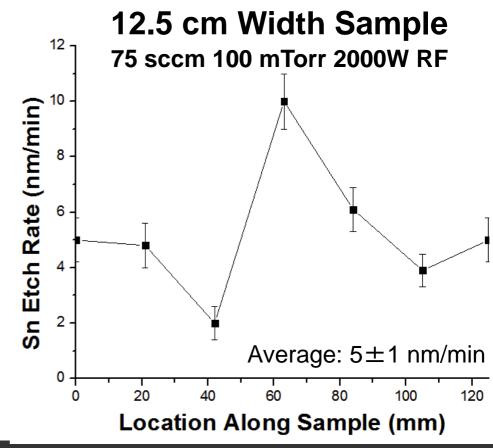
500 W 13.56 MHz **RF-Generator**

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Effect of Sample Size

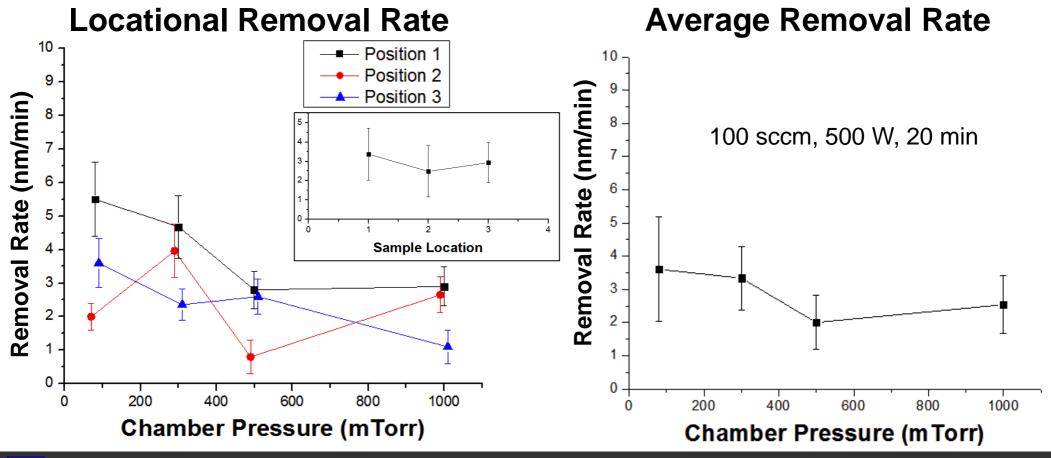
- Small sample etches ~ 7x faster than larger sample
- Unable to etch large sample at same conditions. Why?
 Redeposition
- Higher flow rates required to remove SnH₄ from sample.





Etch Rate Vs. Pressure (100 sccm)

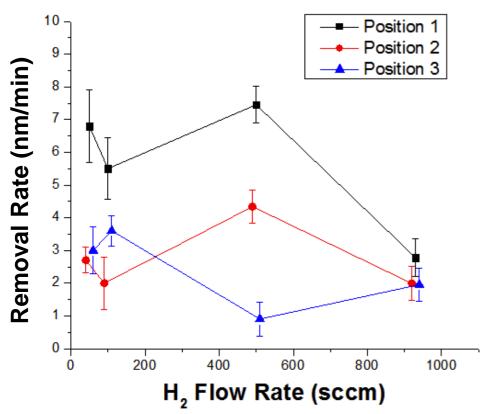
- Etch rate maximized at lowest pressure.
- More likely to redeposit at higher pressures.
- Sample location 2 (furthest from open space) has lowest average Sn removal rate observed.



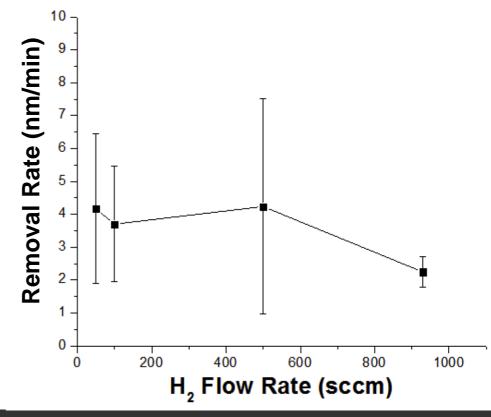
Etch Rate Vs Flow Rate

- Less than 500 sccm, too little flow rate. Above 500 sccm too few radicals.
- 950 sccm flow rate minimizes cross-antenna variation at a cost in removal rate.

Locational Removal Rate



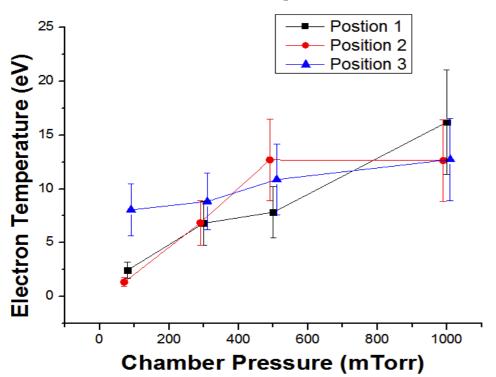
Average Removal Rate



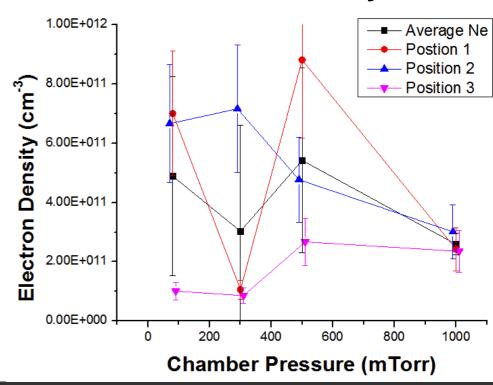
Plasma Parameters Vs. Pressure (100 sccm)

- Increasing chamber pressure results in increasing electron temperature.
- Temperatures appear high and might be the result of beams, which are not perceivable with triple probes.
- Electron density appears independent of pressure.

Electron Temperature

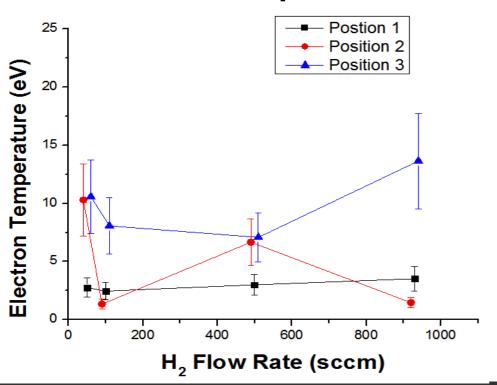


Electron Density

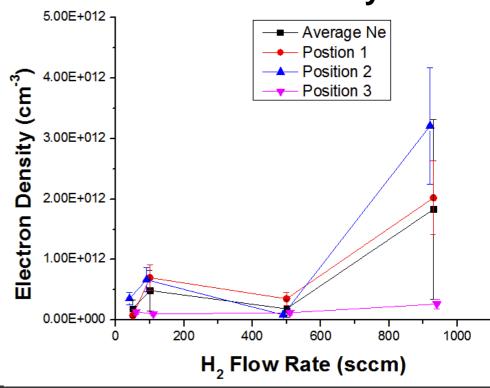


- Electron temperature appears independent of chamber flow rate.
- Electron density is relatively uniform except at 950 sccm where it increases by nearly a factor of 2.

Electron Temperature

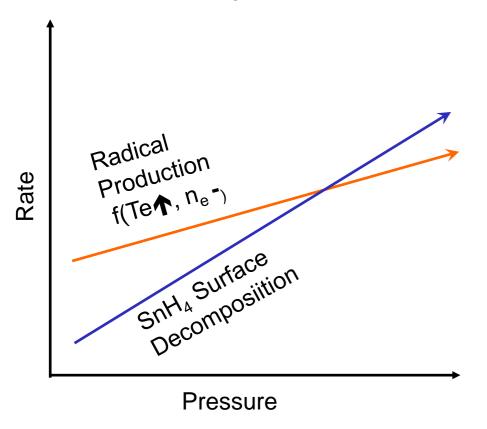


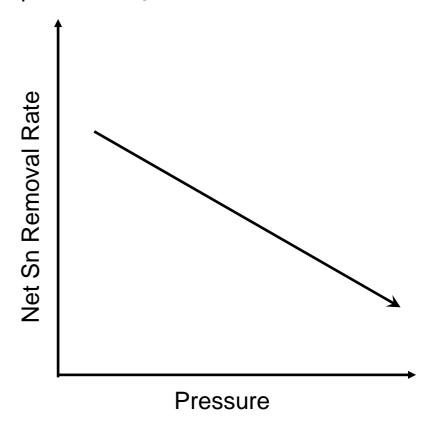
Electron Density



Recap

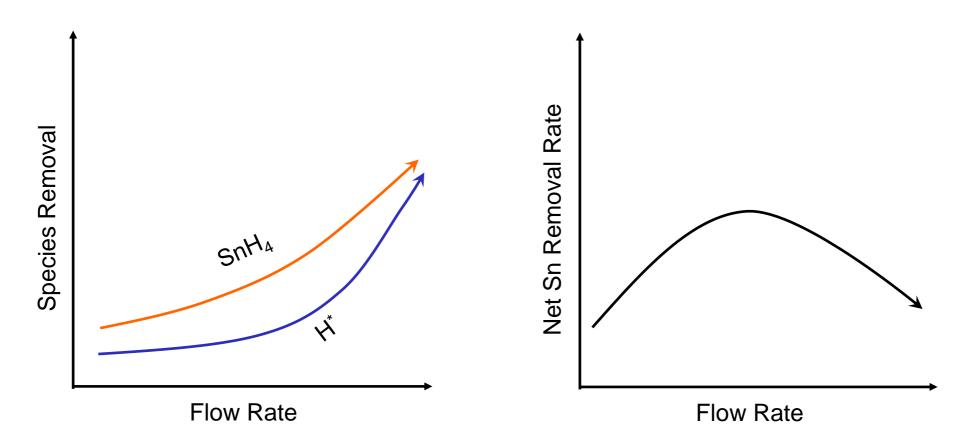
- Increased radical production at higher pressure
 - Should increase etch rate, but removal rate doesn't agree.
- Increased pressure, increases SnH₄ decomposition.





Recap

- SnH₄ removal increases with increasing flow rate
- H* not largely removed until very high flow rates



Best scenario for cleaning is low pressure and the appropriate flow rate.

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Conclusions

- It is possible to remove Sn from a collector optic using an insitu H₂ plasma cleaning system.
- The decomposition of unstable SnH₄ makes removing Sn from large surfaces difficult but not impossible. Sn removal rate ≠ Sn etch rate.
- Higher flow rates are required to remove SnH₄ from larger samples before it decomposes.
- Under presented experimental conditions, optimal Sn removal from large surfaces occurred at 500 sccm, 100 mTorr, 500 W RF power. Net removal rates as high as 7.5 ± 0.5 nm/min.
- Net Sn removal rate less affected by variations in pressure/plasma conditions than by flow rate.

Acknowledgments

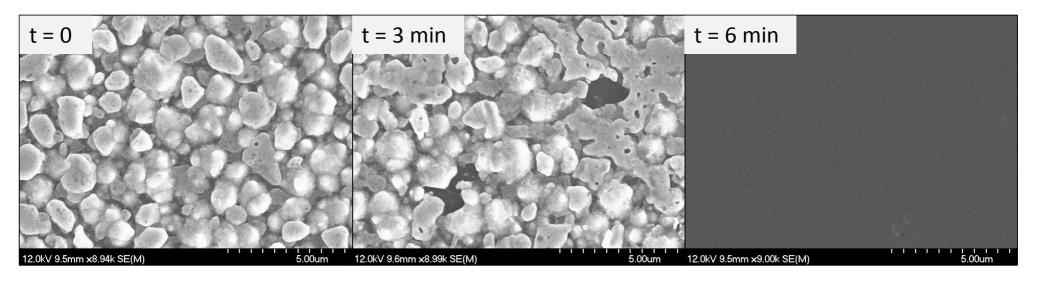
- Special thanks to Cymer, Inc. for funding this research.
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Thank You For Your Attention!

Sample Size Etch Difference

1 cm² Sample Size: Magnetron



125 mm Sample Size: Magnetron

